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Aqua(*n*-hexyl)[3,3'-(propane-1,3-diyldinitrilo)bis(butan-2-one) dioximato- $\kappa^4 N$]cobalt(III) perchlorate

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The title compound, $[Co(C_6H_{13})(C_{11}H_{19}N_4O_2)(H_2O)]ClO_4$, is in the general class of coenzyme B_{12} models which contain a ClO_4 anion and a $[Co(C_6H_{13})(C_{11}H_{19}N_4O_2)(H_2O)]^+$ cation. In the cation, the Co atom has a distorted octahedral coordination, with the *n*-hexyl and H₂O ligands in axial positions. The crystal data reveal some degree of flexibility in the Costa-type system, which is similar to the coenzyme B_{12} .

Comment

Costa-type organocobalt complexes, $[LCo{(DO)(DOH)-pn}R]X$, where *L* is a neutral base, (DO)(DOH)pn is 3,3'-(propane-1,3-diyldinitrilo)bis(butan-2-one)dioximato, *R* is an alkyl group and *X* is an anion (PF₆⁻, ClO₄⁻ *etc.*), exhibit many interesting properties and have been examined in detail as coenzyme B₁₂ models (Randaccio *et al.*, 1989). However, only a few structures of (aqua)alkyl Costa-type complexes ($R = CH_3, X = PF_6^-$, *etc.*) are available (Marzilli *et al.*, 1985), and we report here the structural data for a new compound with $R = n-C_6H_{13}$, (I).



An *ORTEP* plot (Johnson, 1965) of the cation in (I), with the atom-numbering scheme, is shown in Fig. 1. The Co atom exhibits a distorted octahedral stereochemistry and the (DH)(DOH)pn ligand occupies four equatorial positions. The four equatorial N atoms are coplanar within 0.01 (4) Å, and

the Co atom is displaced by 0.033 (1) Å from this mean plane towards the axial alkyl group. The two chemically equivalent halves of the equatorial macrocycle, with the exclusion of C6, are approximately planar. These planes have a dihedral angle of 8.5 (2)° and bend towards the axial water group. The position of the central atom of the propylene bridge appears to be determined mainly by interaction with the axial ligands (Parker *et al.*, 1985). Other bond lengths and angles are in agreement with those in related compounds (Zangrando, Parker, Bresciani-Pahor *et al.*, 1987; Zangrando, Parker & Mezzetti, 1987; Parker *et al.*, 1985).

The oxime H atom takes part in a short intramolecular hydrogen bond linking the oxime O1 and O2 atoms. The H atoms of the water molecule form $O-H\cdots O$ hydrogen bonds with perchlorate O atoms. Details of the hydrogen bonding are given in Table 2.



Figure 1

The molecular structure of the cation of (I) showing 30% probability displacement ellipsoids. H atoms are drawn as spheres of arbitrary radii.

Experimental

Compound (I) was prepared according to the method of Parker *et al.* (1985). A crystal suitable for X-ray diffraction was grown from an acetone–water solution in the dark under aerobic conditions.

Crystal data	
$[Co(C_6H_{13})(C_{11}H_{19}N_4O_2)-$	$D_x = 1.385 \text{ Mg m}^{-3}$
$(H_2O)]ClO_4$	Mo $K\alpha$ radiation
$M_r = 500.86$	Cell parameters from 34
Trigonal, $R\overline{3}$	reflections
a = 42.064 (6) Å	$\theta = 5.25 - 14.77^{\circ}$
b = 42.064 (6) Å	$\mu = 0.87 \text{ mm}^{-1}$
c = 7.052 (3) Å	T = 293 (2) K
$V = 10\ 806\ (5)\ \text{\AA}^3$	Prism, red
Z = 18	$0.8 \times 0.4 \times 0.4$ mm

metal-organic compounds

$$\begin{split} R_{\rm int} &= 0.032\\ \theta_{\rm max} &= 25^\circ\\ h &= -1 \rightarrow 47\\ k &= -50 \rightarrow 1\\ l &= 0 \rightarrow 8 \end{split}$$

3 standard reflections every 97 reflections

+ 52.947P]

 $(\Delta/\sigma)_{\rm max} = 0.001$

 $\Delta \rho_{\rm max} = 0.69 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.41 \text{ e } \text{\AA}^{-3}$

intensity decay: 6.0%

 $w = 1/[\sigma^2(F_o^2) + (0.0828P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

Data collection

Siemens P4 diffractometer
$2\theta/\omega$ scans
Absorption correction: empirical
<i>via</i> ψ scan (North <i>et al.</i> , 1968)
$T_{\min} = 0.687, T_{\max} = 0.707$
4691 measured reflections
4077 independent reflections
2891 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.064$ $wR(F^2) = 0.183$ S = 1.0254077 reflections 298 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Selected geometric parameters (Å, °).

Co1-N1	1.882 (4)	N1-C2	1.299 (7)
Co1-N2	1.919 (4)	N2-C3	1.291 (6)
Co1-N3	1.914 (4)	N2-C5	1.464 (7)
Co1-N4	1.876 (4)	N3-C7	1.455 (7)
Co1-C12	2.023 (5)	N3-C9	1.289 (7)
Co1-O1W	2.116 (4)	N4-O2	1.330 (5)
N1-O1	1.337 (5)	N4-C10	1.298 (7)
N4-Co1-N1	97.5 (2)	N3-Co1-C12	90.8 (2)
N4-Co1-N3	81.31 (19)	N2-Co1-C12	89.3 (2)
N1-Co1-N3	177.83 (17)	N4-Co1-O1W	86.7 (2)
N4-Co1-N2	177.75 (18)	N1-Co1-O1W	87.86 (19)
N1-Co1-N2	81.39 (19)	N3-Co1-O1W	90.23 (18)
N3-Co1-N2	99.68 (18)	N2-Co1-O1W	91.29 (19)
N4-Co1-C12	92.7 (2)	C12-Co1-O1W	178.7 (2)
N1-Co1-C12	91.1 (2)		

The hydrogen-bonding H atoms H1, H1WA and H1WB were located from a difference Fourier map and refined isotropically. The remaining H atoms were treated as riding. The terminal methyl group of the hexyl ligand was found to be disordered over two positions,

Table 2Hydrogen-bonding geometry (Å, °).

D-H	$H \cdots A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$	
1.15 (9) 0.79 (7) 0.75 (9)	1.31 (9) 2.04 (7) 2.20 (9)	2.443 (8) 2.819 (8) 2.940 (10)	167 (8) 171 (8) 166 (8)	
	<i>D</i> -H 1.15 (9) 0.79 (7) 0.75 (9)	$\begin{array}{c c} D-H & H\cdots A \\ \hline 1.15 (9) & 1.31 (9) \\ 0.79 (7) & 2.04 (7) \\ 0.75 (9) & 2.20 (9) \end{array}$	$D-H$ $H \cdots A$ $D \cdots A$ 1.15 (9)1.31 (9)2.443 (8)0.79 (7)2.04 (7)2.819 (8)0.75 (9)2.20 (9)2.940 (10)	

Symmetry code: (i) 1 - x, -y, 1 - z.

C17 and C17'. These two atoms refined to occupancies of 0.31 (5) and 0.69 (5), respectively.

Data collection: *XSCANS* (Siemens, 1994); cell refinement: *XSCANS*; data reduction: *SHELXTL* (Sheldrick, 1995); program(s) used to solve structure: *SHELXTL*; program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BK1502). Services for accessing these data are described at the back of the journal.

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